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A norbornyl route to azasugars: stereoselective synthesis of isofagomine analogues

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Abstract

A stereoselective synthesis of new isofagomine analogues has been achieved from a suitably functionalized cyclopentene intermediate extracted from the norbornyl framework. Double reductive amination or inter- and intramolecular *N*-alkylations are the key steps in constructing the piperidine ring. Isofagomine derivatives exhibit moderate inhibitory activity in enzyme assays. © 2000 Elsevier Science Ltd. All rights reserved.

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Polyhydroxylated piperidine alkaloids and their synthetic analogues have attracted a great deal of attention in recent years due to their ability to mimic sugars and competitively and selectively inhibit glycosidases and glycosyltransferases, the carbohydrate processing enzymes. ¹ These attributes make hydroxylated piperidines (imino- or azasugars) likely therapeutic agents for the treatment of diseases related to metabolic disorders of carbohydrates such as diabetes, cancer, AIDS and viral infections, where glycoprotein processing is crucial. Typical among the natural products that have shown potent glycosidase inhibition are nojirimycin 1,^{1a} deoxynojirimycin 2,^{1a} homonojirimycin 3^{2a} and fagomine 4,^{2b} representing varying levels of oxygenation and stereochemical patterns on the piperidine ring. The promising profile of hydroxylated piperidines has stimulated a search for newer potent analogues based on this ring system. These efforts have largely focused on restructured monosaccharides with one or more additional side arm(s) and either the anomeric oxygen. ring oxygen, or both, being replaced with nitrogen, which, through protonation, could accommodate positive charge in the transition state to improve inhibition. These efforts have led to the design of glycosylamine 5,^{2d} glucosamidine 6,^{1c} isofagomine 7a and b (1-N-iminosugars)³ and even a 1-azafagomine 8, among others, all of which exhibit pronounced glycosidase inhibitory activity. In view of the remarkable inhibition profile of isofagomine 7a,b and current interest in its congeners,³

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we have devised syntheses of homoisofagomine and isofagomine diastereomers and evaluated their activity against glycosidases.

Scheme 1 outlines the retrosynthetic plan for accessing the homoisofagomine **9a** and isofagomine **9b** analogs from the cyclopentene precursor **11**. Oxidative cleavage of the double bond in **11** was expected to deliver **10** to set up either double reductive amination or interand intramolecular *N*-alkylations to give **9** (Scheme 1). The cyclopentene precursor **11** can in turn be accessed from the fragmentation of the norbornyl derivative **12** as reported recently by us.⁴

Scheme 1.

LAH reduction and hydroxyl protection transformed 11 to 13 in high yield. The fully protected olefin 13 was subjected to ozonolysis and the intermediate dialdehyde was directly reduced to furnish diol 14. The diol 14 was readily transformed to the dimesylate 15 to set the stage for interand intramolecular *N*-alkylations to construct the piperidine ring. Exposure of 15 to *p*-toluene-sulphonamide under phase transfer conditions resulted in smooth cyclization to give 16,⁵ (Scheme 2). The *N*-tosyl group in 16 could be removed with sodium naphthalenide and the resulting free amine was characterized as the acetamide 17, which existed as a mixture of two rotamers.^{3e} Deprotection manoeuvres in 17 led to the homoisofagomine 19⁵ via the intermediate 18 (Scheme 2).

To obtain isofagomine, cyclopentene 11 was elaborated as follows. LAH reduction and acetylation furnished 20. OsO₄ mediated catalytic dihydroxylation of 20 gave *cis*-diol 21, essentially as a single diastereomer. Periodate cleavage in 21 led to an intermediate dialdehyde, which was directly subjected to double-reductive amination⁶ to give piperidine 22 as the major product, Scheme 3. Hydrolysis of acetate 22 and oxidation with TPAP yielded aldehyde 23. Decarbonylation in 23 with the Wilkinson's catalyst was smooth and 24 was realized as a single diastereomer. Routine deprotection protocols on 24 delivered isofagomine and its *N*-benzyl derivatives 26⁵ and 25,⁵ respectively (Scheme 3).

Scheme 2. Reagents and conditions: (a) LiAlH₄, THF, 0°C, 30 min, 96%; NaH, BnBr, 0°C \rightarrow rt, overnight, \sim 96%; (b) i. O₃, DCM, -78°C, 5 min, DMS; ii. NaBH₄, EtOH, 6 h, 36% for two steps; (c) MsCl, Et₃N, DCM, -10°C \rightarrow 0°C, >95%; (d) p-TsNH₂, pBu₄N⁺I⁻, KOH, benzene:H₂O (20:1), 20 h, 61%; (e) sodium naphthalenide, -78°C, 1 h; Ac₂O, py, 90%, (f) H₂, Pd/C (10%), EtOH, 18 h, 78%; (g) Et₂O:HCl (3:2), 70°C, 14 h, 93%

Scheme 3. Reagents and conditions: (a) LiAlH₄, THF, 0°C, 30 min, 96%; Ac₂O, DMAP, DCM, 0°C, 45 min, ~90%; (b) OsO₄ (1 mol%), NMMO (50% aq. sol.), Me₂CO:H₂O, 14 h, 84%; (c) i. NaIO₄ (1.3 equiv.), DCM, 0°C, 2 h, ii. BnNH₂, AcOH, NaCNBH₃, MeOH, -10°C \rightarrow rt, 20 h, 49% for two steps; (d). i. KOH, MeOH, 3 h, 92%, ii. "Pr₄NRuO₄, NMMO (97%), mol. sieves 4 Å, 2 h, 78%; (e) Rh(PPh₃)₃Cl, toluene, reflux, 12 h, 60%; (f) 2.5% HCl:Et₂O (1:1), 18 h, >90% for **25**; (g) H₂, Pd/C (10%), EtOH, 18 h, 40%; 2.5% HCl: Et₂O (1:1), 95% for **26**

New isofagomine analogues 19, 25, 26 were assayed for glycosidase inhibition (Table 1). All measurements were carried out with the corresponding nitrophenyl glycoside substrates in aqueous buffer at the appropriate pH. It was surprising to find that homoisofagomine 19 did not inhibit any of the glycosidases. However, 25 was found to be a moderate to strong inhibitor of all the glycosidases used. Its inhibition of α -glucosidase ($K_i = 60 \mu M$) was stronger than of β -glucosidase ($K_i = 1700 \mu M$). The selectivity in inhibition among α - versus β -galactosidases was much poorer. The debenzylated compound 26 was found to be a moderate but selective inhibitor of β -glucosidase. This reversal in selectivities of 25 and 26 for α - and β -glucosidase, respectively, is notable and similar observations have been reported recently for other azasugars. It is also to be noted that isofagomines 7b^{3d} and 26 exhibit very similar β -glucosidase inhibition (see Table 1),

Table 1	
Inhibition constants ^{a,b} (K_i) in μ	ιM

	compound			
Enzyme	19	25	26	7b
α-glucosidase (<i>yeast</i>)	NI	60	NI	NI
β-glucosidase (sweet almond)	NI	1700	140	120
α-galactosidase (green coffee beans)	NI	89	NI	-
β-galactosidase (<i>E. Coli</i>)	NI	180	NI	NI

^aEach 200 μL assay contained indicated enzyme, inhibitor in water(2-3 mM) and nitrophenyl glycosides (2-2.5 mM) in appropriate buffer at optimal temp and pH of each enzyme. ^b Inhibiton constants were determined using Dixon plots of inhibition data.^c No inhibition is observed up to 1mM conc. of inhibitor.

indicating that stereochemical disposition of substituents in isofagomine series has little modulating effect on β -glucosidase inhibition.^{3d}

In conclusion, we have amplified the synthetic utility of the cyclopentanoid building block 11 by devising stereoselective routes to isofagomine analogues. Our preliminary results of enzymatic assays reveal the importance of *N*-substitution in modulating selectivity and inhibition efficacy in azasugars.

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References

- Recent Reviews: (a) Hughes, A. B.; Rudge, A. J. Nat. Prod. Rep. 1994, 135. (b) Jacob, G. S. Curr. Opin. Struct. Biol. 1995, 605. (c) Ganem, B. Acc. Chem. Res. 1996, 29, 340. (d) Bols, M. Acc. Chem. Res. 1998, 31, 1. (e) Heightman, T. D.; Vasella, A. T. Angew. Chem., Int. Ed. 1999, 38, 750. (f) Sears, P.; Wong, C.-H. ibid. 1999, 38, 2300.
- (a) Ikeda, K.; Takahashi, M.; Nishida, M.; Miwa, M.; Kizu, H.; Kameda, Y.; Arisawa, M.; Watson, A. A.; Nash, R. J.; Fleet, G. W. J.; Asano, N. Carbohydr. Res. 2000, 323, 73. (b) Asano, N.; Tomioka, E.; Kizu, H.; Matsui, K. ibid. 1994, 253, 235. (c) Kato, A.; Asano, A.; Kizu, H.; Matsui, K. J. Nat. Prod. 1997, 60, 312. (d) Legler, G. Adv. Carbohydr. Chem. Biochem. 1990, 48, 319
- (a) Ichikawa, Y.; Igarashi, Y., Tetrahedron Lett. 1995, 36, 4585.
 (b) Ichikawa, Y.; Igarashi, Y.; Ichikawa, M.; Suhara, Y. J. Am. Chem. Soc. 1998, 120, 3007.
 (c) Jespersen, T. M.; Dong, W.; Sierks, M. R.; Skrydsurp, T.; Lundt, I.; Bols, M. Angew. Chem., Int. Ed. Engl. 1994, 33, 1778.
 (d) Hansen, A.; Tagmose, T. M.; Bols, M. Chem. Commun. 1996, 2649.
 (e) Ernholt, B. V.; Thomson, I. B.; Jensen, K.; Bols, M. Synlett 1999, 6, 701.
 (f) Hensen, S. U.; Bols, M. J. Chem. Soc., Perkin Trans. 1 2000, 911 and references cited therein.

- 4. Mehta, G.; Mohal, N. Tetrahedron Lett. 1999, 40, 5791.
- 5. All new compounds reported here were racemic and were characterized on the basis of their spectral data (¹H and ¹³C NMR, IR, Mass). Selected spectral data (1 H NMR, 13 C NMR): 16: $δ_{H}$ (300 MHz, CDCl₃): 7.62 (2H, d, J = 8.1Hz), 7.36-7.31 (5H, m, Ar–H), 7.28 (2H, d, J=8.1 Hz), 4.5 (2H, ABq, J=12 Hz), 4.18 (1H, dd, J=5.7, 6.0 Hz), 3.85 (1H, dd as t, J = 5.1 Hz), 3.64 - 3.55 (2H, m), 3.41 (1H, dd, J = 5.4, 12 Hz), 3.08 (1H, dd, J = 3.6, 12.0 Hz), 2.98(1H, dd, J = 6.3, 12.3 Hz), 2.88 (1H, dd, J = 6.3, 12.3 Hz), 2.41 (3H, s), 2.19-2.10 (1H, m), 1.83-1.62 (2H, series of the serm), 1.31 (3H, s), 1.29 (3H, s); δ_C (75 MHz, CDCl₃): 143.47 (C), 138.35 (C), 133.94 (C), 129.62 (CH, 2C), 128.37 (CH, 2C), 127.62 (CH, 2C), 127.56 (CH), 127.52 (CH, 2C), 108.87 (C), 76.54 (CH), 72.94 (CH₂), 70.79 (CH), 63.58 MHz, D₂O): 4.01 (1H, br. s), 3.56–3.44 (3H, series of m), 3.34–3.25 (2H, series of m), 3.04 (1H, d, J = 13.5 Hz), 2.68 (1H, dd as t, J = 13 Hz), 2.09–1.99 (1H, m), 1.88–1.76 (1H, m), 1.40–1.28 (1H, m); δ_C (75 MHz, D₂O): 71.68 (CH), 66.12 (CH), 59.93 (CH), 48.80 (CH₂), 47.13 (CH₂), 32.81 (CH), 31.98 (CH₂). **25**: δ_H (300 MHz, D₂O): 7.38 (5H, m), 4.24 $(1H, \frac{1}{2}ABq, J = 13.3 Hz)$, 4.15 $(1H, \frac{1}{2}ABq, J = 13.3 Hz)$, 3.97 (1H, br. s), 3.32–3.28 (3H, m), 3.01 $(1H, \frac{1}{2}ABq, \frac{1}{2}ABq$ d, J = 13 Hz), 2.67 (1H, dd as t, J = 13 Hz), 2.13–2.07 (1H, m), 0.86 (3H, d, J = 6.6 Hz); δ_C (75 MHz, D₂O): 132.05 (CH, 2C), 130.96 (CH), 129.96 (CH, 2C), 128.90 (C), 73.10 (CH), 66.58 (CH), 60.79 (CH₂), 57.04 (CH₂), 56.38 (CH_2) , 30.53 (CH), 14.49 (CH_3) . **26**: δ_H (300 MHz, D_2O): 3.99 (1H, br. s), 3.33 (1H, dd, J = 2.7, 10.5 Hz), 3.29–3.15 (2H, series of m), 3.02 (1H, d, J=13 Hz), 2.59 (1H, dd as t, J=12.6 Hz), 2.10-2.0 (1H, m), 0.88 (3H, d, J=6.6 Hz); $\delta_{\rm C}$ (75 MHz, D₂O): 73.40 (CH), 66.16 (CH), 49.11 (CH₂), 48.80 (CH₂), 30.26 (CH), 14.60 (CH₃).
- 6. Baxter, E. W.; Reitz, A. B. J. Org. Chem. 1994, 59, 3175.
- 7. Butters, T. D.; van den Broek, L. A. G. M.; Fleet, G. W. J.; Krulle, T. M.; Wormald, M. R.; Dwek, R. A.; Platt, F. M. Tetrahedron: Asymmetry 2000, 11, 113.